Ca²⁺-mobilising properties of synthetic fluoro-analogues of *myo*-inositol 1,4,5-trisphosphate and their interaction with *myo*-inositol 1,4,5-trisphosphate 3-kinase and 5-phosphatase

Stephen T. Safrany¹, Deborah Sawyer², Richard J.H. Wojcikiewicz¹, Stefan R. Nahorski¹ and Barry V.L. Potter²

Departments of ¹Pharmacology and ²Chemistry, University of Leicester, Leicester, LEI 7RH, UK

Received 23 August 1990

The ability of two fluoro-analogues of D-myo-inositol 1,4,5-trisphosphate (Ins(1,4,5)P₃) to mobilize intracellular Ca²⁺ stores in SH-SY5Y neuro-blastoma cells has been investigated. DL-2-deoxy-2-fluoro-scyllo-Ins(1,4,5)P₃ (2F-Ins(1,4,5)P₃) and DL-2,2-difluoro-2-deoxy-myo-Ins(1,4,5)P₃ (2,2-F₂-Ins(1,4,5)P₃) were full agonists (EC₅₀s 0.77 and 0.41 μ M respectively) and slightly less potent than D-Ins(1,4,5)P₃ (EC₅₀ 0.13 μ M), indicating that the axial 2-hydroxyl group of Ins(1,4,5)P₃ is relatively unimportant in receptor binding and stimulation of Ca²⁺ release. Both analogues mobilized Ca²⁺ with broadly similar kinetics and were substrates for Ins(1,4,5)P₃ 3-kinase but, qualitatively, were slightly poorer than Ins(1,4,5)P₃. 2F-Ins(1,4,5)P₃ was a weak substrate for Ins(1,4,5)P₃ 5-phosphatase but 2,2-F₂-Ins(1,4,5)P₃ was apparently not hydrolysed by this enzyme, although it inhibited its activity potently ($K_i = 26 \mu$ M).

Second messenger; Inositol phosphate; Fluoro-analogue; Ca2+ mobilisation

1. INTRODUCTION

D-myo-inositol 1,4,5-trisphosphate (Fig. 1, (1)) is a second messenger which mediates the release of Ca²⁺ from intracellular stores [1,2] via a receptor which has been cloned and sequenced [3] and which, when reconstituted, mediates Ca²⁺ release in response to Ins(1,4,5)P₃ [4]. A major challenge is now the elucidation of the structural basis for interaction of Ins(1,4,5)P₃ both with its receptor and with the metabolic enzymes, Ins(1,4,5)P₃ 3-kinase and 5-phosphatase, and the rational chemical design of agonists, antagonists and enzyme inhibitors. Recent progress in inositol phosphate chemistry [5,6] and molecular recognition has been reviewed [7].

Several inositol ring-modified and phosphate-modified analogues have been synthesized [5,6] and some progress has been made to understand the role of the 3 phosphate and hydroxyl groups of Ins(1,4,5)P₃ in receptor binding specificity and stimulation. Isosteric replacement of a hydroxyl group with fluorine [8] has led to fluorinated *myo*-inositol [9,10] and *myo*-inositol phosphate analogues [11,12]. D-3-fluoro-3-deoxy-*myo*-inositol inhibits cell growth in NIH 3T3 cells [9] and 5-fluoro-5-deoxy-*myo*-inositol is taken up by L1210 cells and incorporated into cellular phospholipid [10]. Three reports of biological activity for ring-modified

Correspondence and present address: B.V.L. Potter, School of Pharmacy and Pharmacology, University of Bath, Claverton Down, Bath, BA2 7AY, UK

analogues of Ins(1,4,5)P₃, including DL-2-deoxy-Ins(1,4,5)P₃ (Fig. 1 (2)), have appeared [13,14,15], but no biological evaluation of fluorinated inositol phosphates has yet been reported.

We report here a study of the interaction of the two synthetic analogues 2-deoxy-2-fluoro-scyllo-inositol 1,4,5-trisphosphate (2-F-Ins(1,4,5)P₃) (Fig. 1, (3)) and 2,2-difluoro-2-deoxy-myo-inositol 1,4,5-trisphosphate (2,2-F₂-Ins(1,4,5)P₃) (Fig. 1, (4)) with the Ca²⁺-releasing receptor of SH-SY5Y neuroblastoma cells and the metabolic enzymes Ins(1,4,5)P₃ 5-phosphatase and 3-kinase.

2. MATERIALS AND METHODS

2.1. Synthetic analogues

DL-2-deoxy-2-fluoro-scyllo-inositol 1,4,5-trisphosphate (Fig. 1, (3)) and DL-2,2-difluoro-2-deoxy-myo-inositol 1,4,5-trisphosphate (Fig. 1, (4)) were synthesized from protected myo-inositol precursors by respective diethylaminosulphur trifluoride fluorination of 1-prop-1-enyl-3,6-di-O-benzyl-4,5-O-isopropylidene-myo-inositol [16] and 1-allyl-3,6-di-O-benzyl-4,5-O-isopropylidene-2-myo-inosose [16], removal of non-benzylic protecting groups, followed by bis(2-cyanoethyl)-N,N-diisopropylamino-phosphine phosphitylation, oxidation of the resulting trisphosphites with t-butyl hydroperoxide and deprotection using sodium in liquid ammonia, as described for Ins(1,4,5)P₃ [17]. Analogues were purified on an ion-exchange column of DEAE Sephadex A-25 using a gradient of triethylammonium bicarbonate buffers, pH 8.0, and were isolated as the triethylammonium salt and assayed by quantitative phosphate analysis. Full chemical details of the synthetic procedures will appear elsewhere.

2.2. Cell culture, permeabilisation and Ca2+ release

Monolayers of SH-SY5Y human neuroblastoma cells (passage 75-95), initially a kind gift of Dr J.L. Biedler (Sloane-Kettering In-

Fig. 1. Structures of synthetic analogues of Ins(1,4,5)P₃. Only D-isomers are shown.

stitute, New York, USA), were grown as described [18]. Prior to permeabilisation, the cells were harvested in 10 mM Hepes, 0.9% NaCl pH 7.4, containing 0.02% EDTA, electroporated [20] and $^{45}\text{Ca}^{2+}$ release experiments were performed as described [19]. The temporal characteristics of Ca^{2+} mobilisation from electroporated SH-SY5Y cells, maintained in the same buffer as for $^{45}\text{Ca}^{2+}$ experiments, were monitored using a Ca^{2+} -specific electrode [21]. All cell culture reagents were from GIBCO Ltd, D-Ins(1,4,5)P₃ was from Calbiochem, $^{45}\text{CaCl}_2$ (approx. 1000 Ci/mmol) was obtained from Amersham plc, ATP, quin-2 and oligomycin were obtained from Sigma. EC₅₀ values were derived using ALLFIT computer-assisted curve fitting [22]. Combined data from a number of independent experiments (n) are expressed as mean \pm SEM, where $n \ge 3$.

2.3. Ins(1,4,5)P₃ 5-phosphatase activity

Human erythrocyte ghosts (7 mg protein/ml) were prepared according to Downes et al. [23] and stored at -40° C. Protein concentration was determined as described [25]. Ins(1,4,5)P₃ or fluoroanalogues (100 μ M) were incubated at 37°C for 60 min in the presence of ghosts (5 mg protein/ml) or inactivated (boiled) ghosts in a buffer consisting of 30 mM Hepes, 2 mM MgCl₂, pH 7.2. Incubations were terminated by boiling and Ins(1,4,5)P₃ and its analogues were then assayed for ability to release Ca2+ from permeabilised SH-SY5Y $5[^{32}P]$ -Ins(1,4,5)P₃ Inhibition of metabolism 2,2-F₂-Ins(1,4,5)P₃ was performed essentially as described for inositol 1,4,5-trisphosphorothioate [24]. Erythrocyte ghosts (1 mg protein/ml) were incubated at 37°C for 12 min in the presence of 30 μ M $Ins(1,4,5)P_3$ (approx. 5000 dpm [32P]- $Ins(1,4,5)P_3$) and 1-100 μM 2,2-F₂-Ins(1,4,5)P₃. Under these conditions no more than 20% of the substrate was consumed.

2.4. $Ins(1,4,5)P_3$ 3-kinase activity

Whole rat brains were minced, homogenised, diluted to a concentration of 20% w/v in 150 mM sucrose and centrifuged at 100 000 × g for 90 min. The resulting supernatant was stored at $-40^{\circ}\mathrm{C}$ until required. Ins(1,4,5)P₃ or fluoro-analogues (1 mM) were incubated at 37°C for 60 min in the presence of this crude 3-kinase preparation (5% w/v) or inactivated (boiled) enzyme preparation in a buffer consisting of 50 mM Tris-maleate, 20 mM MgCl₂, 10 mM Na₂ATP, 5 mM 2,3-bisphosphoglycerate, 0.1% BSA, pH 7.5. Incubations were terminated by boiling and the mixtures were assayed for ability to release $^{45}\mathrm{Ca}^{2+}$ from SH-SY5Y cells.

3. RESULTS AND DISCUSSION

Myo-inositol possesses 5 equatorial hydroxyl groups and one axial hydroxyl group. Using fluorinated $Ins(1,4,5)P_3$ analogues we have attempted to probe the role of the unique axial 2-hydroxyl group in determining the affinity and specificity of $Ins(1,4,5)P_3$ for its receptor and metabolic enzymes. The EC_{50} of

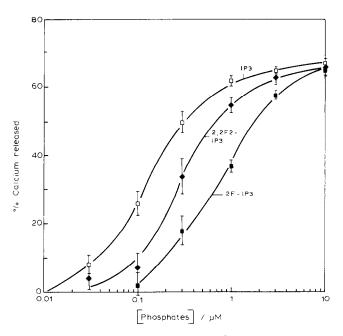


Fig. 2. Dose-response curves for release of Ca²⁺ by Ins(1,4,5)P₃ and analogues in SH-SY5Y neuroblastoma cells. For experimental details see section 2.

Ins(1,4,5)P₃-induced Ca²⁺ release in permeabilised SH-SY5Y human neuroblastoma cells was 0.13 μ M (Fig. 2, Table I). For DL-2,2-F₂-Ins(1,4,5)P₃ (Fig. 1, (4)) and DL-2F-Ins(1,4,5)P₃ (Fig. 1, (3)) the EC₅₀ values were 0.41 and 0.77 μ M respectively (Fig. 2, Table I). Both analogues were full agonists for Ca²⁺ mobilisation (Fig. 2) and released Ca²⁺ with kinetics similar to Ins(1,4,5)P₃, although re-uptake of Ca²⁺ was slower for the analogues than for Ins(1,4,5)P₃ (Fig. 3).

These data suggest that the analogues have slightly lower affinity than Ins(1,4,5)P₃ for the receptor. This has also been confirmed by radioligand binding to the rat cerebellar Ins(1,4,5)P₃ receptor (A.L. Willcocks, S.R. Nahorski and B.V.L. Potter, data not shown). It should be noted, however, that the synthetic fluorocompounds are racemic mixtures. L-Ins(1,4,5)P₃ has previously been shown to be essentially inactive at binding to and inducing Ca²⁺ release at the stereospecific

Table I Ca2+release1 Fold shifts in EC504 EC_{50} (μ M) 3-kinase 5-phosphatase (n = 12)treatment2 treatment3 (n = 3-4)(n=3)D-Ins(1,4,5)P₃ 0.13 ± 0.01 24 ± 6.5 10 ± 3.4 DL-2F-Ins(1,4,5)P3 0.77 ± 0.08 1.5 ± 0.9 1.2 ± 0.3 $DL-2,2-F_2-Ins(1,4,5)P_3 0.41 \pm 0.05$ 7 \pm 3.0 0.2 ± 0.3

⁴ Fold shifd is defined by ([EC₅₀(after)/EC₅₀(before)] -1)

¹ ⁴⁵Ca²⁺ release from permeabilised SH-SY5Y neuroblastoma cells ² After 1 h treatment of Ins(1,4,5)P₃ or analogue (1 mM) with 5% 3-kinase preparation

 $^{^3}$ After 1 h treatment of Ins(1,4,5)P3 or analogue (100 $\mu M)$ with 5 mg protein/ml 5-phosphatase at 37°C

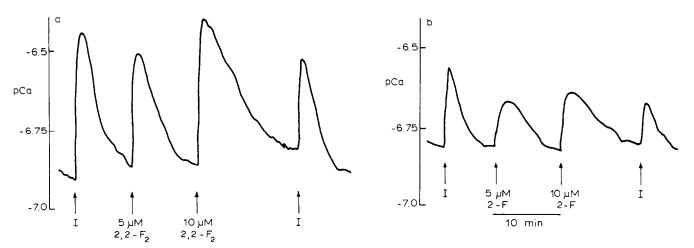


Fig. 3. Kinetics of Ca^{2+} release induced by $Ins(1,4,5)P_3$ and synthetic fluoro-analogues monitored in electrically permeabilised SH-SY5Y neuroblastoma cells using a Ca^{2+} -specific electrode. Suspensions of permeabilised cells (3-4 mg protein/ml) were challenged with 5 μ M D-Ins(1,4,5)P₃ (I) and (a) 5 μ M and 10 μ M 2,2-F₂-Ins(1,4,5)P₃ or (b) 5 μ M and 10 μ M 2F-Ins(1,4,5)P₃. Data shown are representative of results from 3 similar independent experiments.

Ins(1,4,5)P₃ receptor [7]. Thus, the true EC₅₀ values for D-2,2-F₂-Ins(1,4,5)P₃ and D-2F-Ins(1,4,5)P₃ may be taken as 0.21 μ M and 0.39 μ M respectively and fluorosubstitution at the 2-position therefore makes the analogues only some 1.6- and 3-fold less potent than Ins(1,4,5)P₃.

These data should be compared with an EC₅₀ of 6 μ M for D-6-deoxy-Ins(1,4,5)P₃-induced Ca²⁺ release in SH-SY5Y cells (J. Strupish, S. Safrany, D. Dubreuil, S.D. Gero, S.R. Nahorski and B.V.L. Potter, unpublished observations) and with an EC₅₀ of 0.5 μ M reported for DL-2-deoxy-Ins(1,4,5)P₃-induced Ca²⁺ release (i.e. $0.25 \mu M$ for the D-isomer compared with 0.2 μ M for D-Ins(1,4,5)P₃) in permeabilised macrophages [14]. Qualitatively, therefore, there is little difference between the 2-deoxy and 2-fluorinated analogues and since the EC50 values are only slightly higher than Ins(1,4,5)P₃ itself, the axial 2-hydroxyl, in contrast to the 6-hydroxyl group, cannot be making a major contribution to receptor binding and activation. This is also borne out by the tolerance of the Ca²⁺ mobilising receptor for certain bulky substitutions at the 2-position [14,15].

The three phosphate groups of $Ins(1,4,5)P_3$ will most likely make ionic interactions with positively charged centres of the $Ins(1,4,5)P_3$ receptor and make the major contribution to binding energy. The three hydroxyl groups at the 2, 3, and 6 positions will be either hydrogen bond donors to the protein, or acceptors of hydrogen bonds from it. Complete deletion of single or multiple hydroxyl groups can lead to drastic loss of Ca²⁺-mobilising ability, exemplified by low of D-6-deoxy-Ins(1,4,5)P₃ potency DL-1,2,4-cyclohexane trisphosphate [13], but apparently not for 2-deoxy-Ins $(1,4,5)P_3$ (Fig. 1, (2)) [14]. Isosteric and isoelectronic replacement of a hydroxyl group by fluorine [8], however, might be predicted to

have a less drastic effect, provided H-bonding interactions could still be made with the protein. 2,2-F₂-Ins(1,4,5)P₃ and 2F-Ins(1,4,5)P₃ differ structurally only by the fact that the former possesses an axial fluorine atom rather than a proton and is therefore unable to donate a hydrogen bond. 2,2-F₂-Ins(1,4,5)P₃ is clearly a more potent analogue than 2F-Ins(1,4,5)P₃, presumably since the axial fluorine atom is able to make a more favourable interaction with the receptor and, indeed, this analogue is only marginally (1.6-fold) less potent than Ins(1,4,5)P₃ itself. We propose, therefore, that the 2-hydroxyl group of Ins(1,4,5)P₃ accepts a hydrogen bond from the receptor

Ins(1,4,5)P₃ is metabolised by two major routes: phosphorylation by a 3-kinase and dephosphorylation by a 5-phosphatase [7]. Treatment of solutions of $Ins(1,4,5)P_3$ or fluoro-analogues with a 3-kinase preparation, followed by assay of Ca²⁺-releasing ability, gave shifts in EC₅₀ on account of 3-position phosphorylation as shown (Table I). Clearly, both fluoro-analogues are weaker substrates $Ins(1,4,5)P_3$ for the 3-kinase, but $2,2-F_2-Ins(1,4,5)P_3$ appeared to be significantly better than 2F-Ins(1,4,5)P₃, presumably since the axial fluorine atom of the former acts more successfully to mimic the normal electronic environment at the 2-position adjacent to the crucial 3-hydroxyl group. Whilst not an absolute requirement for activity, the 2-hydroxyl group is clearly an important element in the recognition of substrates by $Ins(1,4,5)P_3$ 3-kinase.

We have also investigated the interaction of the fluoro-analogues with $Ins(1,4,5)P_3$ 5-phosphatase from human erythrocyte ghosts. Treatment of solutions of $Ins(1,4,5)P_3$ or fluoro-analogues with a 5-phosphatase preparation, followed by assay of Ca^{2+} -releasing ability, gave shifts in EC_{50} , on account of dephosphorylation as shown (Table I). Thus, whilst $2F-Ins(1,4,5)P_3$

appears to be a weaker substrate for 5-phosphatase than Ins(1,4,5)P₃, there was no significant shift in EC₅₀ value for 2,2-F₂-Ins(1,4,5)P₃. This is surprising, on account of the known low specificity of 5-phosphatase for Moreover, study substrates [7,13].a inhibition $2,2-F_2-Ins(1,4,5)P_3-induced$ of $5[^{32}P]Ins(1,4,5)P_3$ breakdown by erythrocyte 5-phosphatase (data not shown; method similar to [24]) showed that this analogue binds to the enzyme with high affinity $(K_i = 26 \pm 4.1 \ \mu\text{M}; \text{ cf. } K_m \text{ for } D$ $Ins(1,4,5)P_3$, 40 μM [24]). While $Ins(1,4,5)P_3$ 5-phosphatase is specific for the D-enantiomer of Ins(1,4,5)P₃, the analogue is a racemate and this unexpected result may indicate that L-2,2-F₂-Ins(1,4,5)P₃ has a considerably higher affinity for 5-phosphatase than would be expected (cf. K_i for L-Ins(1,4,5)P₃, 124 and is inhibiting breakdown of μ M [24]) D-2,2- F_2 -Ins(1,4,5) P_3 . This seems plausible in the light of data showing that a 2-arylacyl-substituted L-Ins(1,4,5)P₃ analogue was a highly potent inhibitor of 5-phosphatase with a K_i of 3.8 μ M [15]. Further analysis must await resolution of the pure enantiomers of $2,2-F_2-Ins(1,4,5)P_3$. The slow hydrolysis of $2F_2-Ins(1,4,5)P_3$. $Ins(1,4,5)P_3$ is also in accord with the observation that 2-deoxy-Ins(1,4,5)P₃ is hydrolysed more slowly than Ins(1,4,5)P₃ by both erythrocyte ghost and brain cytosol 5-phosphatase [14]. Our results support the notion that the 2-hydroxyl group of Ins(1,4,5)P₃ is not absolutely required for activity, but it may be involved in substrate recognition by $Ins(1,4,5)P_3$ 5-phosphatase.

The metabolism of the fluoro-analogues was examined using electrically-permeabilised SH-SY5Y cells, which exhibit both Ins(1,4,5)P₃ 5-phosphatase and 3-kinase activities [26]. 2,2-F₂-Ins(1,4,5)P₃ and 2F-Ins(1,4,5)P₃ gave similar Ca^{2+} -release profiles to that induced by Ins(1,4,5)P₃, but Ca^{2+} re-uptake, which parallels inositol phosphate metabolism, was retarded (fig. 3); e.g. for 5 μ M D-Ins(1,4,5)P₃ the time taken for Ca^{2+} concentration to decay to 50% of the maximal response was on average 2.8 min, whereas for 10 μ M DL-2,2-F₂-Ins(1,4,5)P₃ the time taken was 5.4 min. This is consistent with a combination of the effects of 5-phosphatase and 3-kinase already demonstrated.

We conclude that both analogues are recognized by all three $Ins(1,4,5)P_3$ binding proteins and that the 2-hydroxyl group of $Ins(1,4,5)P_3$ does not play an absolute role either in receptor binding and activation, or enzyme activity, although it may be an important recognition element.

Acknowledgements: This work was supported by SERC (Molecular Recognition Initiative) and the Wellcome Trust. We thank the University of Leicester for a Research Scholarship (to D.S.). B.V.L.P. is a Lister Institute Fellow.

REFERENCES

- [1] Berridge, M.J. (1987) Annu. Rev. Biochem. 56, 159-193.
- [2] Berridge, M.J. and Irvine, R.F. (1989) Nature 341, 197-205.
- [3] Furuichi, T., Yoshikawa, S., Miyawaki, A., Wada, K., Maeda, N. and Mikoshiba, K. (1989) Nature (Lond.) 342, 32-38.
- [4] Ferris, C.D., Huganir, R.L., Supattapone, S. and Snyder, S.H. (1989) Nature 342, 87-89.
- [5] Billington, D.C. (1989) Chem. Soc. Rev. 18, 83-122.
- [6] Potter, B.V.L. (1990) Nat. Prod. Rep. 7, 1-24.
- [7] Nahorski, S.R. and Potter, B.V.L. (1989) Trends Pharmacol. Sci. 10, 139-144.
- [8] Schlosser, M. (1978) Tetrahedron 34, 3-17.
- [9] Kozikowski, A.P., Fauq, A.H., Powis, G. and Melder, D.C. (1990) J. Amer. Chem. Soc. 112, 4528-4531.
- [10] Moyer, J.D., Reizes, O., Surender, A., Jiang, C., Malinowski, N. and Baker, D.C. (1988) Mol. Pharmacol. 33, 683-689.
- [11] Maracek, J.F. and Prestwich, G.D. (1989) Tetrahedron Lett. 30, 5401-5404.
- [12] Ley, S.V., Parra, M., Redgrave, A.J., Sternfeld, F. and Vidal, A. (1989) Tetrahedron Lett. 30, 3557-3560.
- [13] Polokoff, M.A., Bencen, G.H., Vacca, J.P., de Solms, J., Young, S.D. and Huff, J.R. (1988) J. Biol. Chem. 263, 11922-11927.
- [14] Hirata, M., Watanabe, Y., Ishimatsu, T., Ikebe, T., Kimura, Y., Yamaguchi, K., Ozaki, S. and Koga, T. (1990) J. Biol. Chem. 264, 20303-20308.
- [15] Hirata, M., Yanaga, F., Koga, T., Ogasawara, T., Watanabe, Y. and Ozaki, S. (1990) J. Biol. Chem. 265, 8404-8407.
- [16] Gigg, J., Gigg, R., Payne, S. and Conant, R. (1987) J. Chem. Soc. Perkin Trans I 1757-1762; 2411-2414.
- [17] Cooke, A.M., Gigg, R. and Potter, B.V.L. Tetrahedron Lett. (1987) 28, 2305-2308.
- [18] Lambert, D.G., Ghataorre, A.S. and Nahorski, S.R. (1989) Eur. J. Pharmacol. 165, 71-77.
- [19] Strupish, J., Cooke, A.M., Potter, B.V.L., Gigg, R. and Nahor-ski, S.R. (1988) Biochem. J. 253, 901-905.
- [20] Wojcikiewicz, R.J.H., Lambert, D.G. and Nahorski, S.R. (1990) J. Neurochem. 54, 676-685.
- [21] Clapper, D.L. and Lee, H.C. (1985) Biochem. J. 260,
- 13947-13954. [22] DeLean, A., Munson, P.J. and Rodbard, D. (1978) Am. J.
- Physiol. 235, E97-E102. [23] Downes, C.P., Mussat, M.C. and Michell, R.H. (1982)
- Biochem. J. 203, 169-177. [24] Cooke, A.M., Nahorski, S.R. and Potter, B.V.L. (1989) FEBS Lett. 242, 373-377.
- [25] Bradford, M.M. (1976) Anal. Biochem. 72, 248-254.
- [26] Wojcikiewicz, R.J.H., Cooke, A.M., Potter, B.V.L. and Nahorski, S.R. (1990) Eur. J. Biochem., in press.